|  |  |
| --- | --- |
| 1. | 2. |
| 3. | 4. |
| 5. | 6. |
| 7. | 8. |
| 9. | 10. |
| 11. | 12. |

Did you hear the joke about cobalt, radon, and yttrium?

\_ \_ \_ \_ \_ \_ \_ \_ \_ \_ \_ \_ .

7 10 5 2 8 12 3 8 11 1 9 6

SI CHEM 1311- Chapter 1 Review

Describe the key differences between physical and chemical properties of a substance.

Fill in the blanks with the SI metric unit for each measurement.

Volume \_\_\_\_\_\_\_\_\_\_\_\_\_

Mass \_\_\_\_\_\_\_\_\_\_\_\_\_

Length \_\_\_\_\_\_\_\_\_\_\_\_\_

Temperature \_\_\_\_\_\_\_\_\_\_\_\_

Amount of substance \_\_\_\_\_\_\_\_\_\_\_\_

Which of the following masses could be weighed on a scale that ranges from 5-150kg?

1. 2.3x105 g
2. 2.3x104 g
3. 2.3x103 g

The density of copper is 8.96 g/cm3. What is the mass of a cube of copper that is approximately 8 cm on each side?

An object has a mass of 0.00190330g. How many significant numbers does this digit have?

Describe the movement of molecules in a solid substance? In a gas?

Solve the following problem using correct significant figures.

(1.304 + 24.36) ÷ (3.006 x 0.00238)

An exothermic reaction reported a change in temperature of 18.3oC. What is the change in temperature in Kelvins?

Which of the following is an example of a chemical change?

1. Compression of a gas
2. Evaporation of a liquid
3. Combustion of a gas
4. Condensation of a vapor

Which of the following is an example of a pure substance?

1. Air
2. Baking soda
3. Ocean water
4. Juice concentrate

Round the number off to three significant figures: 0.00850043 \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Round the number off to four significant figures: 2,456,900 \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

A scientist is using propane and finds that it has a density of 36.28lb/ft3. He needs to report it in metric units. What is the density in kg/m3? (1 lb = 454g, 1 in = 2.54cm)